

Review Article

# **Edaravone: A Review on Analytical Method and its Determination in Biological Matrix and Synthetic Mixture**

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#### **ABSTRACT**

Edaravone is a potent free radical scavenger (antioxidant) mainly use in the form of injection. It is used in the treatment of various cardiovascular diseases like acute ischemic stroke as well as in gastrointestinal injuries. This review article represent the various analytical methods which has been reported for estimation of edaravone in biological matrix as well as in synthetic mixture. The spectrophotometric techniques like fluorescent assay and ratio derivative spectroscopy; Chromatographic methods like HPLC, HPTLC and RP HPLC were reported.

Keywords: Edaravone, analytical methods, Compendial Method, UV spectroscopic method

# INTROCUTION<sup>[1-2]</sup>

Edaravone is 5-methyl-2-phenyl-2,4-dihydro-3H-pyrazol-3-one is apyrazole derivative appears as white to off white crystalline powder. The drug is freely soluble in Distilled Water. solubility in water is 3 g/1 L. Edaravone is a weak base with pKa values of 7<sup>(3)</sup>, Fivemembered Pyrazole Ring. Edaravone melts at 127-131 °C. Boils at 287° C.

Me N N N Edaravone

Alternative Name: MCI-186 Chemical Name: 3-Methyl-1-phenyl-2-pyrazolin-5-one

M.Wt: 174.2 Formula: C10H10N2O Figure: 1 Structure of edaravone

### MECHANISM OF ACTION [3]

Edaravone has been reported to antioxidant effects because it can quench hydroxyl radicals and hydroxyl radicaldependent lipid peroxidation. Edaravone reduces elevated levels of hydroxyl radicals and superoxide radicals in several models of ischemia. In early studies of antioxidant activity of edaravone, its pKa was found to be 7.0, and the rate of oxidation for edaravone was positively correlated with pH. The putative mechanism underlying the antioxidant action of edaravone is electron transfer from an edaravone anion to peroxyl radical, and this reaction breaks the chain oxidation of lipids.

**How to cite this article:** DA Patel, R Hasumati, R Patel; Edaravone: A Review on Analytical Method and its Determination in Biological Matrix and Synthetic Mixture; PharmaTutor; 2014; 2(10); 80-84



Figure: 2 Mechanism of Edaravone

Edaravone is excreted as unmetabolized drug ( $^{\sim}1\%$ ) or metabolized by sulfation (5–13%) or glucuronidation (68–83%) and excreted in urine within 24hours of administration.

Edaravone is a neuroprotective agent used for aiding neurological recovery following acute brain ischemia and subsequent cerebral infarction. [4] It acts as a potent antioxidant and strongly scavenges free radicals, protecting against oxidative stress and neuronal apoptosis. [5-7] Edaravone has been shown to attenuate methamphetamine- and 6-OHDAinduced dopaminergic neurotoxicity in the striatum and substantia nigra, and does not affect methamphetamine-induced dopamine release or hyperthermia. [8,9] It has also been demonstrated to protect against MPTPmediated dopaminergic neurotoxicity to the substantia nigra, though notably not to the striatum. [10-12]

# Combination of edaravone<sup>[13]</sup>

Edaravone+ozagrel
Edaravone+alteplase (tPA)
Edaravone +citicholine sodium

# Marketed formulation of edaravone<sup>[13]</sup>

Radicut®, Radicut bag

### 1. Analytical Method

A. Compendial Method: Edaravone is not official in Pharmacopoeia.

# B. Reported Method:

I. Fluorescent Method: A Novel Fluorescent Assay for Edaravone with Aqueous Functional Cdse Quantum Dots.



ISSN: 2347-7881

Table No.1: Summary of Fluorescent methodfor edaravone [14]

| Drug      | Method            | Quantum Dots            | Calibration range |
|-----------|-------------------|-------------------------|-------------------|
| Edaravone | Fluorescent Assay | Aqueous Functional Cdse | 1.45–17.42 μg/mL  |

## II. Chromatographic Methods:

The high-pressure liquid chromatography (HPLC) for residue determination. HPTLC methods are widely used chromatographic methods in the analysis of Edaravone in plasma. RP HPLC method also developed for determination of concentration of edaravone in human serum and also for simultaneous determination of edaravone and citicoline sodium.

Table No.2: Summary of Chromatographic Method of Edaravone

| Title              | Method   | Mobile phase                             | Stationary                | Wave   | Reference |
|--------------------|----------|--|---------------------------|--------|-----------|
|                    |          |  | phase                     | Length |           |
| Invitro estimation | RP-HPTLC | -  | Pre coated                | -      | 15        |
| of Edaravonein     |          |  | RP-18 GF <sub>254-</sub>  |        |           |
| Human Plasma       |          |  | aluminum                  |        |           |
|                    |          |  | sheet                     |        |           |
| Determination of   | HPLC     | 0.05mol/L ammonium                       | Diamonsil C <sub>18</sub> | 233nm  | 16        |
| phenyl hydrazine   |          | acetate - acetonitrile                   | column                    |        |           |
| Residue in         |          | (80:20)                                  |                           |        |           |
| edaravone          |          |  |                           |        |           |
| Determination of   | HPLC     | 1%acetic acid                            | Hypersil -                | 243 nm | 17        |
| edaravone and its  |          | :methanol                                | ODC <sub>18</sub>         |        |           |
| related substance  |          | (40:60)                                  | column                    |        |           |
| Estimate Conc. of  | RP HPLC  | H <sub>3</sub> PO <sub>4</sub> :Methanol | Hypersil C <sub>18</sub>  |        | 18        |
| edaravone in       |          | (50:50)                                  | column                    | 240 nm |           |
| human serum        |          |  |                           |        |           |

# III. UV spectroscopic method

First order derivative spectroscopy and Ratio derivative spectroscopic technique was developed for simultaneous determination of edaravone and citicolin sodium.

The ratio derivative spectroscopy method is based on dividing the spectrum for a mixture in to standard spectra for each of analysis and to obtain a spectrum that is independent of analyte concentration used as devisor.

Table No.3:Summary of UV spectroscopic method

| Title  | Method                         | Zero crossing point for edaravone | Zero crossing point for citicoline sodium | R <sup>2</sup> | REF. |
|--|--------------------------------|-----------------------------------|---|----------------|------|
| Simultaneous estimation of                           | Ratio derivative spectroscopic | 258.40                            | 267 nm                                    | 0.999          | 19   |
| Edaravone and citicoline sodium in synthetic mixture | method                         |                                   |   | 0.999          |      |



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| Simultaneous         | First order   | 245.60 nm | 271.20 nm | 0.9996 | 19 |
|----------------------|---------------|-----------|-----------|--------|----|
| estimation of        | derivative    |           |           |        |    |
| Edaravone and        | spectroscopic |           |           | 0.9996 |    |
| citicoline sodium in | method        |           |           |        |    |
| synthetic mixture    |               |           |           |        |    |

#### Table No.4:RP HPLC Method for simultaneous estimation of edaravone and citicoline sodium

| Title                | Method  | Mobile       | Stationary phase | Wave   | Ref. |
|----------------------|---------|--------------|------------------|--------|------|
|                      |         | phase        |                  | length |      |
| Simultaneous         | RP HPLC | Acetonitrile | Phenomenexluna®  | 244    | 20   |
| estimation of        |         | and water    |                  | nm     |      |
| edaravone and        |         | (70:30)      |                  |        |      |
| citicoline sodium in |         |              |                  |        |      |
| synthetic mixture    |         |              |                  |        |      |

#### DISCUSSION

Presented systematic review covers the current analytical methods for the determination of Edaravone and its combination in pharmaceutical and biological samples like serum and plasma. HPLC method were found to be most widely use for edaravone. Various chromatographic conditions are presented in table.

#### CONCLUSION

The sensitivity, specificity, and better separation efficiency enable HPLC to be used frequently for simultaneous qualitative and quantitative determination of edaravone. The presented information is useful for the future study for researcher involved in formulation development and quality control of edaravone.

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